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Rockbridge Extension Master Gardeners

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Water Quality and Conservation



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Welcome to 'Water Quality and Conservation'

Read Chapter 17, in your Master Gardener Handbook before viewing these slides.

- Browse the Suggested Readings at the end of these slides. They contain online sources that will be helpful for your learning.
- The Test Your Knowledge section is for fun and review



Welcome to *What Do You Know About Water Quality?* This is an interactive program that will help you learn about Virginia's water resources, how water quality is measured and what the indicators mean.



What I Will Learn in This Module (Objectives)

1. Description of run-off and leaching
2. Point and non-point sources of pollution
3. Water quality indicators
4. Chemical pesticides and their impact on water quality
5. Guidelines for proper management of chemicals used in and around the home to improve the quality of water supply



INTRODUCTION

We all need clean water. We need it not only for personal use, but also for:



[Photo credit:](#)



[Photo credit](#)

Manufacturing



[Photo credit](#)

Transportation



[Photo credit](#)

Wildlife



© Merle Ann Loman

[Photo credit](#)

Treating wastes



[Photo credit](#)

Food Production



and many other uses.





Photo credit

Water quality refers to the condition of the water. How water is used determines what the condition needs to be.



[Photo credit](#)

[Photo credit](#)



Drinking water must be in better condition (cleaner) than water used for swimming and fishing.



However, water used for swimming and fishing must be clean enough to protect the health of people and fish.



Regardless of how water is used, good water quality is a must. However, nearly everything we do affects the quality of water.

Vehicles, industry and natural events create air-borne particles and gases that enter waterways with dust, rain and snow.



Traffic Northern VA.
[Photo credit](#)



Sewage treatment adds nutrients and other pollutants to waterways.

Agriculture and yard care produces sediment, nutrient and pesticide pollution.



[Photo credit](#)

Many types of chemicals can enter rivers, lakes and groundwater through improper use and storage.

Motor oil and other automotive chemicals can enter waterways from parking lots and highways.



There are many sources of water pollution. Pollution that originates from a specific place is called **POINT SOURCE POLLUTION**. This includes outflows from sewage plants, factories, landfills, feed lots and underground storage tanks.



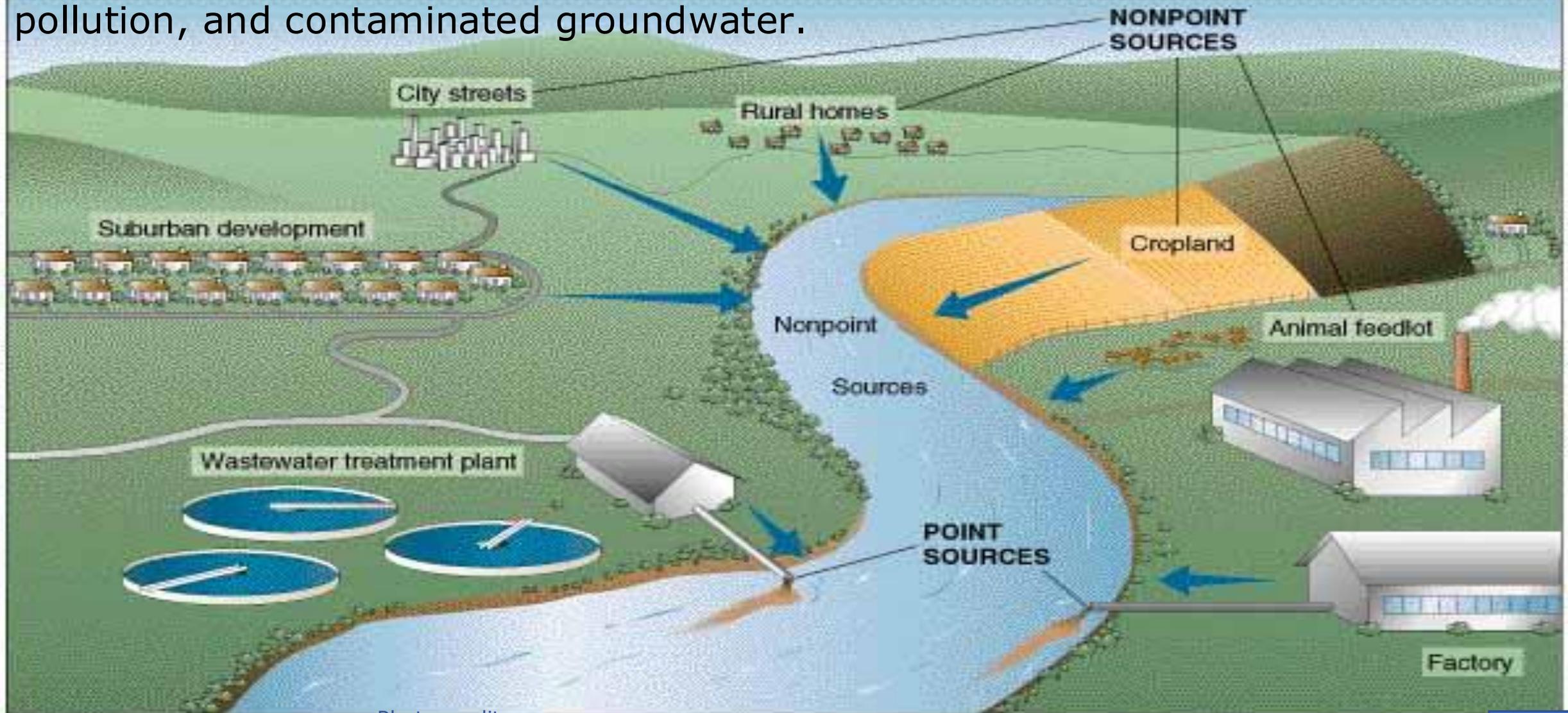
[Photo credit](#)



[Feedlot Photo credit](#)



Pollution that comes from a wide area is called **NON-POINT SOURCE POLLUTION**. This includes runoff from parking lots, roads, farms, yards, stream banks, air pollution, and contaminated groundwater.



[Photo credit](#)



Chesapeake Bay

[Photo credit](#)

Water pollution can also originate from natural sources. Mineral-rich soils, natural oil seepage, coastal saltwater, wildlife and natural disasters can all affect water quality.



There are many types of water pollution. Some of these include:

- Petroleum Products
- Nutrients
- Sewage
- Solid Wastes
- Pesticides
- Toxic Wastes
- Bacteria
- Sediments
- Heated Water
- ... And others



The most widespread types of pollution are sediment, nutrients and bacteria.

Erosion moves soil into waterways. The soil creates suspended sediment that blocks sunlight, smothers aquatic life and adds contaminants.



Primary plant nutrients include nitrogen, phosphorus and other elements. Excess nutrients increase algal growth. The algae die and the resulting decay consumes oxygen.



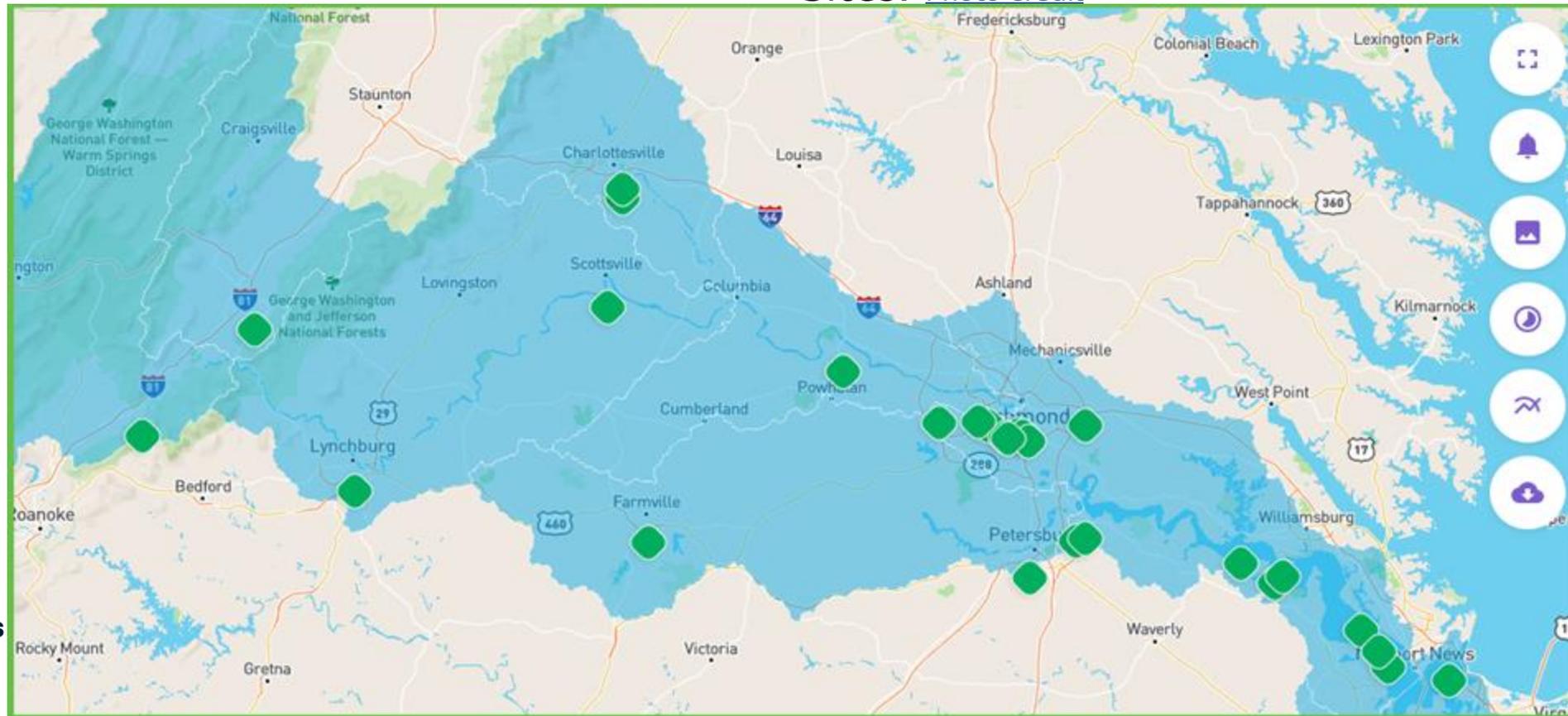
Shenandoah Riverkeeper identified 73 herds of cattle with direct access to the North Fork, South Fork, and mainstream Shenandoah River in 2018

Partially treated sewage, farm runoff and animal wastes are major sources of nutrients and bacteria.



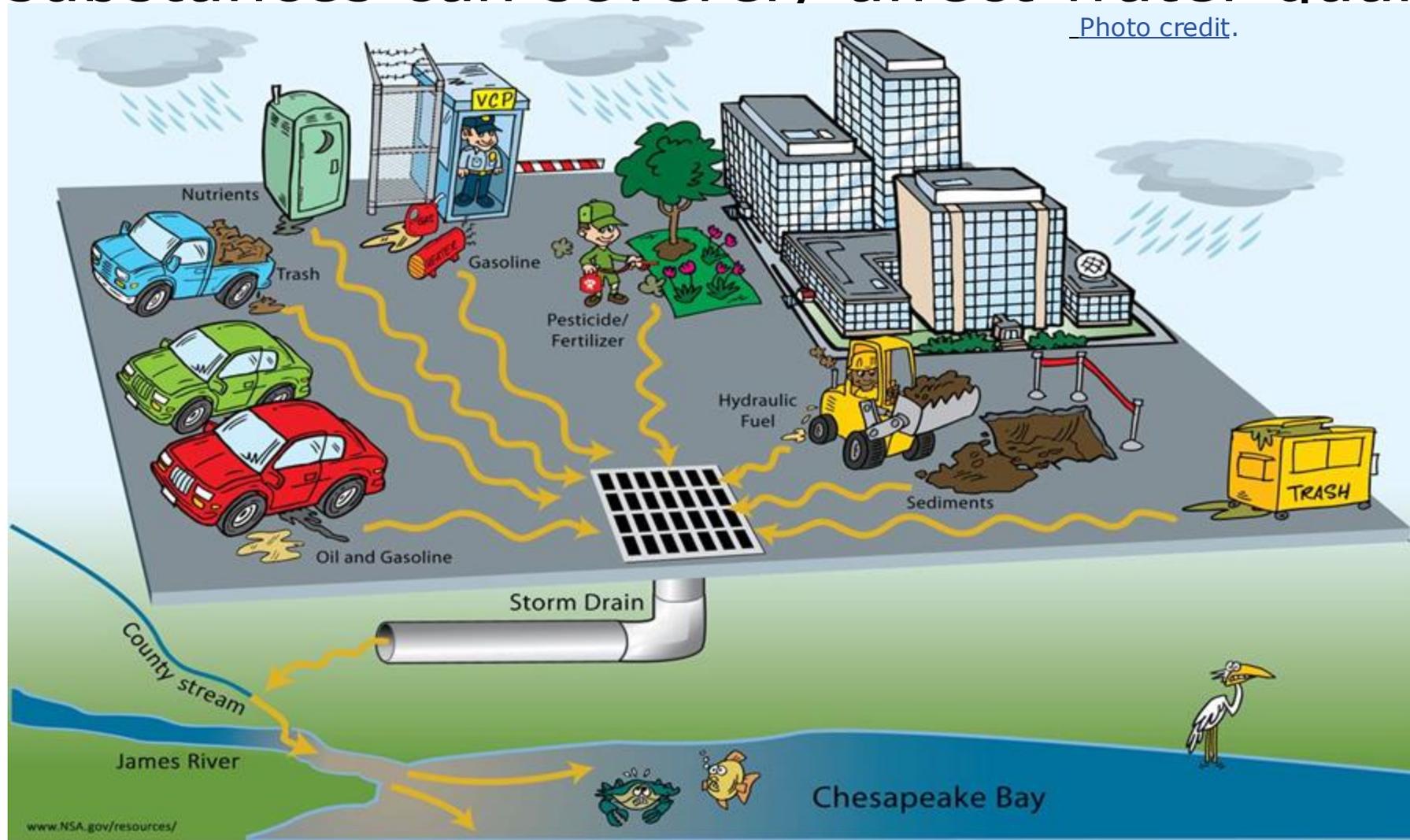
Bacterial contamination is becoming a chronic problem in many waterways. The most prevalent sources are poorly treated sewage and wildlife, pet and livestock waste.

James River Water Quality Monitoring Sites. [Photo credit](#)



In addition, petroleum products, farm and garden chemicals, heavy metals and other toxic substances can severely affect water quality.

[Photo credit.](#)



An important concern is the long-term exposure to low levels of these pollutants. Cancer is a possible risk. However, these substances may also affect hormone and immune systems.



There are many chemical measures used to determine water quality. This program will help you learn about important water quality indicators.



[Photo credit.](#)



Water Quality Indicators

- pH
- Phosphate
- Salinity
- Sediment
- Temperature
- Toxic Chemicals
- Pesticides
- Dissolved and suspended solids
- Turbidity
- Ammonia
- Bacteria
- Hardness
- Nitrogen

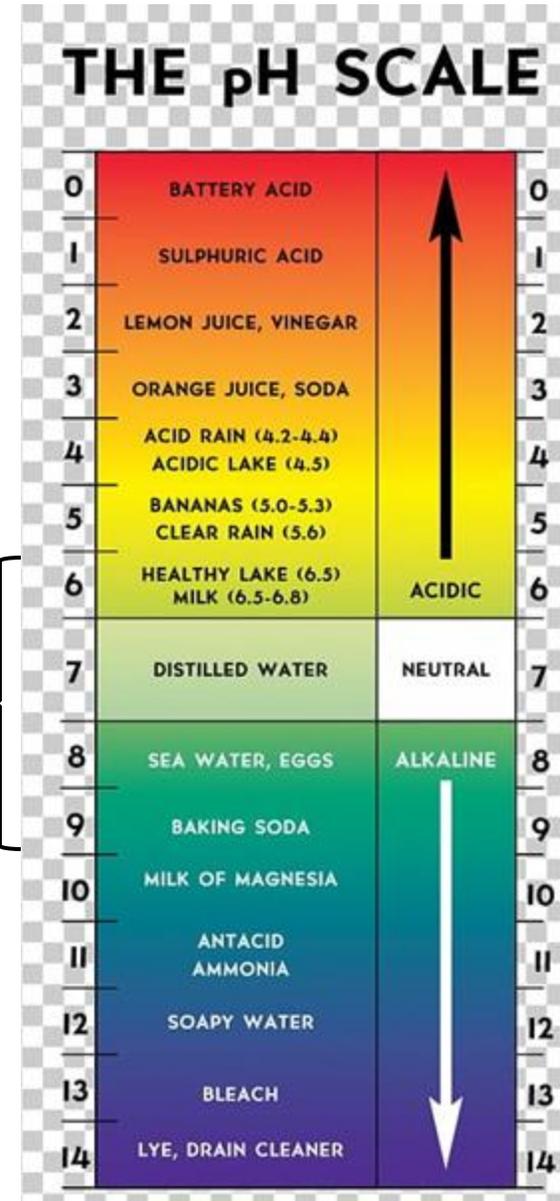


pH

pH is the measure of acidity vs. alkalinity.

The pH scale is zero to 14. Seven is neutral, below seven is acidic, and above seven is basic or alkaline.

Most aquatic organisms exist within a pH range of 5.5 to 9.5.



PHOSPHATE

- Phosphate is an important plant nutrient used for root growth
- Phosphate binds to soil particles and is not usually water soluble
- Excess phosphate can cause algae blooms and reduce water quality
- Chicken and swine production are a major source of phosphate pollution

Discharge
Pipes

Hog farm in
North Carolina.
[Photo credit](#)



SALINITY

Salinity refers to the salt concentration in water, mostly sodium chloride

Salinity is measured in parts per thousand (ppt) or grams per liter

Ocean water has a salinity of about 35 ppt



Dredging at Virginia Beach after a hurricane, 2019, was halted due to sea turtle migration. [Photo credit](#)





SEDIMENT

Sediment is the “conglomerate of materials, organic and inorganic, that can be carried away by water, wind or ice. Erosion causes loose soil to enter waterways.” Sediment can be carried downstream by water flow

(Photo Credit: NASA Visible Earth, via USGS)

Suspended sediment blocks sunlight and reduces dissolved oxygen.

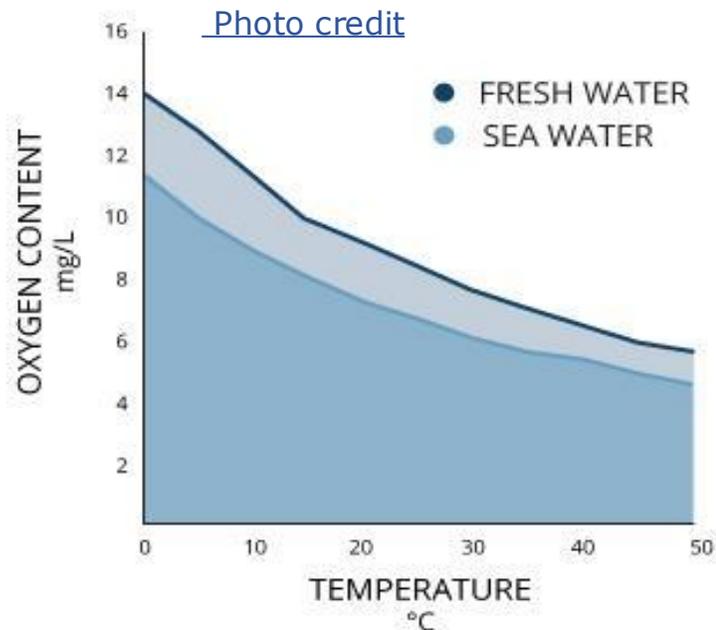
Heavier sediment particles quickly settle out of the water.



TEMPERATURE

Temperature affects the oxygen-carrying capacity of water. Dissolved oxygen concentrations are affected by diffusion and aeration, photosynthesis, respiration and decomposition

Rapid temperature change and temperature extremes can stress aquatic organisms. Dissolved oxygen depletion is the most common cause of fish kills



As the water warms, the amount of dissolved oxygen decreases.



Hypoxic, so called 'Dead Zones' are areas where the oxygen concentration is so low that animals can suffocate and die. The largest hypoxic zone in the U.S. and the second largest worldwide forms in the northern Gulf of Mexico adjacent to the Mississippi River from runoff from farms and cities that drain into the Mississippi River.

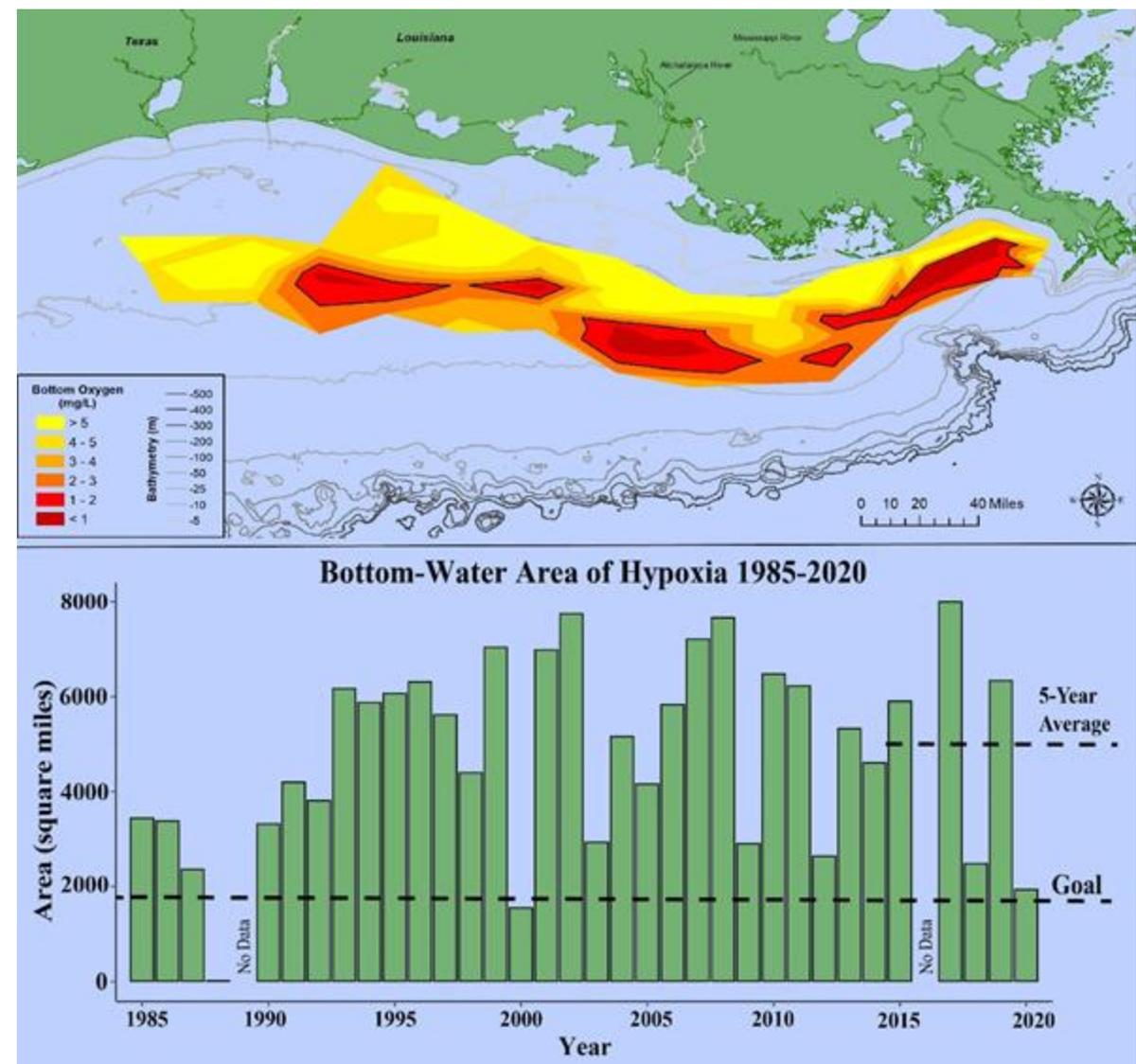


Photo credit



TOXIC CHEMICALS

Toxic chemicals usually come from industry and energy production.

The effects are often not known until years after they have entered the environment.

Toxic chemicals include heavy metals (lead, mercury), organic compounds (DDT, PCB), inorganic substances (arsenic) and others.

Many common household chemicals are toxic to aquatic life (cleaning fluids, paints, thinners, polishes, etc.).



PESTICIDES

Pesticides are rated by a number of factors:

- How well they adhere to soil
- How water soluble they are
- How long they last in the environment
- How they affect organisms

Some types of pesticides include:

Herbicides	Insecticides	Fungicides
Algicides	Miticides	Rodenticides



DISSOLVED SOLIDS

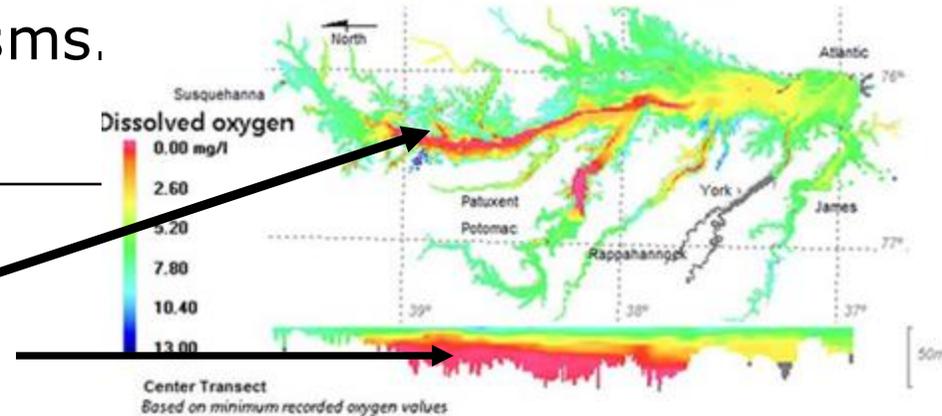
Natural sources of dissolved solids in groundwater include bedrock and salt deposits; additional sources include on-site sewage systems, runoff and wastewater from urban, industrial, or agricultural areas, runoff from roads and wastewater from extraction activities such as coal mining and natural gas drilling.

Suspended sediment and dissolved substances reduce the amount of oxygen water can carry.

As suspended sediments settle, they can smother bottom (benthic) organisms.

A large part of the Chesapeake Bay has low dissolved oxygen.

Percentage of Water in Maryland's Mainstem Chesapeake
Chesapeake Bay Dissolved Oxygen
First August 2018 Cruise - Aug 1, 2018-Aug 15, 2018



[Photo credit](#)

[Total Dissolved Liquids in Household Water](#)

*Data from the first August cruise:
August 1-15, 2018. Courtesy MD
DNR*



TURBIDITY

- Turbidity refers to water clarity
- Sediments suspended in the water increase turbidity
- Clay and silt particles produce most of the turbidity
- Turbidity is related to hardness
- High turbidity prevents light from reaching aquatic life



Big Otter River at U.S. Route 460 in Bedford County, Virginia, June 15, 2013. [Photo credit](#)



AMMONIA

- Ammonia is produced by the decay of protein matter and animal waste.
- Ammonia is toxic to most aquatic life, especially at high pH.
- Bacteria readily convert ammonia to nitrate (a plant nutrient).
- Ammonia is a form of nitrogen.



BACTERIA

- Most bacteria are important in nutrient and other biological cycles.
- Excess nutrients cause algal blooms. As algae die and decay, the high bacterial load rapidly consumes dissolved oxygen.



Fish kill due to low dissolved oxygen.

[Photo credit](#)



BACTERIA

Certain types of bacteria indicate animal and human waste pollution.

Escherichia coli are coliform bacteria found in the intestines of warm-blooded organisms. Most strains are harmless but one *E. coli* strain can cause severe diarrhea and kidney damage.



HARDNESS

Hardness refers to the concentration of calcium and magnesium in water.

Hard water has high concentrations of these elements.
Soft water has low concentrations.

Water hardness often originates from limestone.
Excess hardness causes quality problems.



NITROGEN

Nitrogen makes up 80% of the atmosphere

Nitrogen is a primary plant nutrient

Nitrogen is water soluble and moves easily from surface to groundwater



Photo credit

Excess nitrogen causes algal blooms that reduce water quality



Photo credit:
usgs.gov



Nitrogen Cycle

Under normal conditions, the nitrogen cycle keeps the amount of available nitrogen in balance with the demands.

However, excessive use of fertilizers and nutrient-rich sewage release have created a surplus of nitrate in waterways.

The result has been excess algae and bacteria with reduced dissolved oxygen.

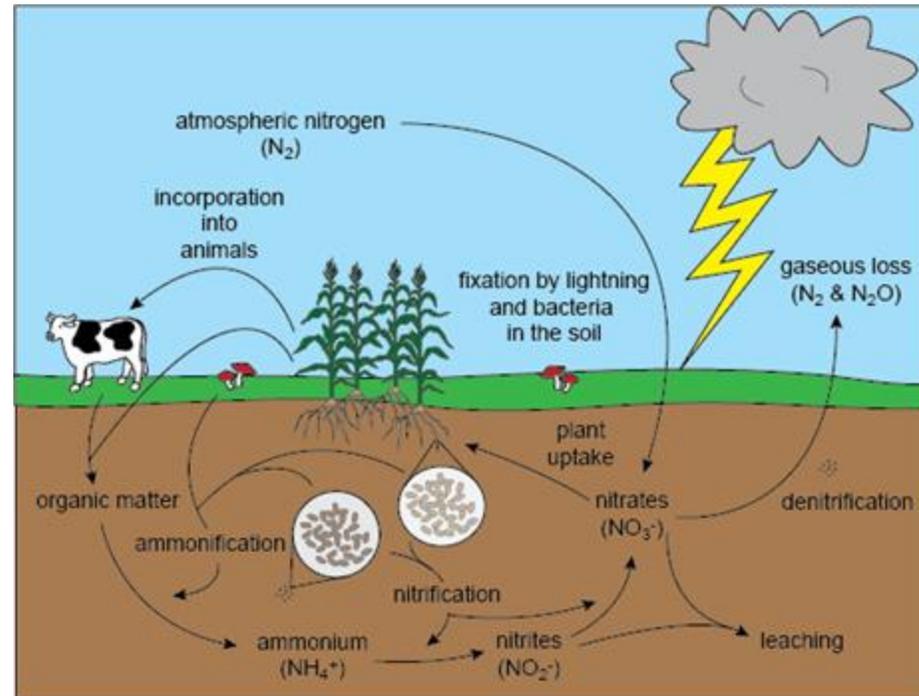


Photo credit



Valuable Water Quality Web Sites (Click on bars to go to sites)

[USGA Water Information Pages](#)

[US Environmental Protection Agency](#)

[Chesapeake Bay Program](#)



PARTS PER MILLION

Most dissolved substances found in water are measured in parts per million (ppm) or even smaller amounts.

This means that for every one million parts (units) of water there is a certain number of parts of the substance.

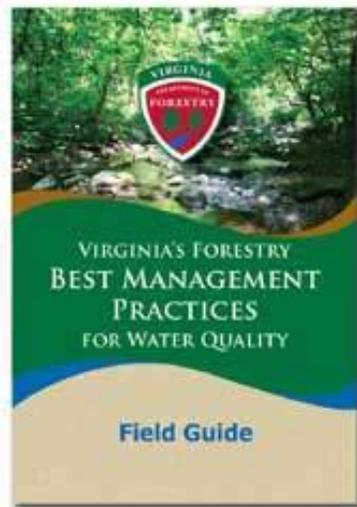


Virginia's Anti-Degradation Policy

Virginia has a constitutional mandate to protect the quality of its natural resources (Virginia Constitution, Art. XI, Sec. 1). In keeping with this mandate the state has adopted an anti-degradation policy which requires the protection of existing high quality surface waters (lakes, rivers, streams, etc.) and groundwater. The policy also provides for the restoration of all other waters of the state to a condition that would permit all reasonable uses (VA. Code Sec. 62.1-44.4(2)). The Virginia Water Control Board is responsible for setting groundwater quality standards.



You can help by adopting effective Best Management Practices for your home, farm, school and community.



Click on image to find this document. Photo credit



Apple Orchard Falls, Sedalia.
Photo credit



Steps in Designing a Water-wise Landscape

- Group plants with similar watering needs together (hydrozones)
- Limit lawns to the appropriate size for your needs (replace some lawn with trees, shrubs, perennials)
- Limit areas with high irrigation needs
- Reduce runoff and erosion
- Provide water-wise maintenance and irrigation practices



Suggested Readings

[A Primer on Water Quality](#)

[Groundwater Quality](#)

[Home Water Quality](#)

[Pesticides and Aquatic Animals: A Guide to Reducing Impacts on Aquatic Systems](#)

[Virginia's major waterways](#)



Test Your Knowledge

[Crossword
Puzzle](#)

[Help Desk Quiz](#)

[Apply What
You Have
Learned](#)



Apply What You Have Learned

1. Identify one potential source of water pollution on your property
2. Develop a hydrozone in your yard / garden
3. Have the water going into your house tested for hardness and pH

Click here to
return to
'[Test your
Knowledge](#)'



Help Desk Quiz

Answers on next slide

1. What steps can I take to reduce the amount of water I use in my gardens and flower beds?
1. How much should I water my vegetable garden?
1. Can I use "gray" water (or household waste water) for my garden plants?

Click here
to return to
'[Test Your Knowledge](#)'



Help Desk Quiz

1. What steps can I take to reduce the amount of water I use in my gardens and flower beds?

Answer: Group plants together by their water needs; limit the amount of lawn you have (plant trees with large mulch areas around them or perennial flower beds that require less water); reduce runoff; use drip irrigation to water plants; if plants will tolerate shade, plant them in the shade to reduce water requirements; use mulch and ground cover; add organic matter to planting areas

2. How much should I water my vegetable garden?

Answer: One inch per week either from rain or irrigation.

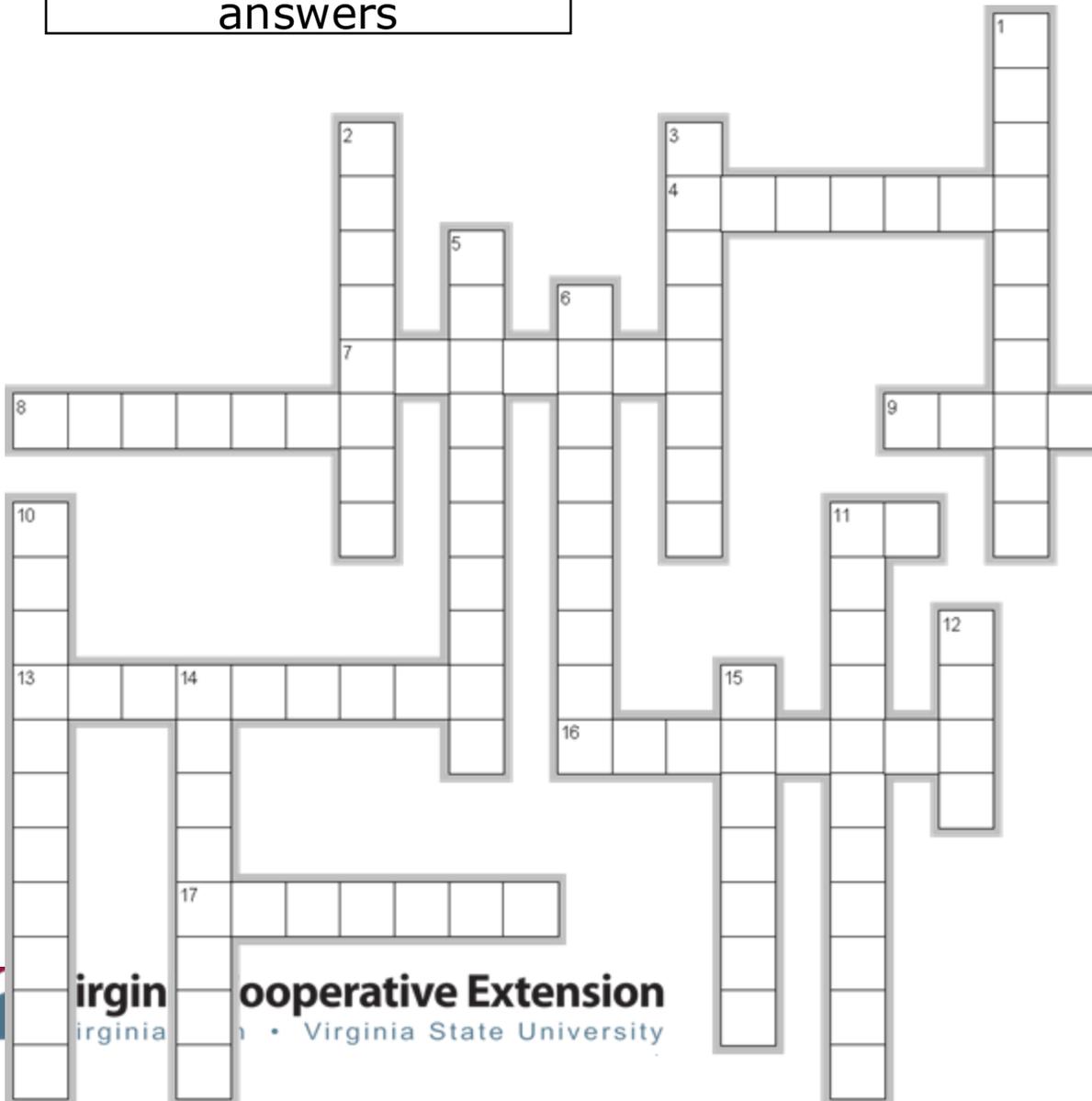
3. Can I use "gray" water (or household waste water) for my garden plants?

Answer: Yes, but with care. No more than about ½ gallon per square foot per week. The greatest danger is from the salts in soaps.



Crossword Puzzle
Next Slide has
answers

Answers



ACROSS

- 4. Product of protein decay
- 7. Soil on the move
- 8. Groundwater layer
- 9. pH less than 7
- 11. Power of the Hydrogen
- 13. Important in root growth
- 16. Particles of organic matter
- 17. Primary plant nutrient

DOWN

- 1. Buffers pH
- 2. What coliform is
- 3. Measures calcium and magnesium
- 5. Urban runoff
- 6. What oxygen is in water
- 10. Measured in °F and °C
- 11. Pollution from a pipe
- 12. pH greater than 7
- 14. Saltwater measure
- 15. From living matter

[Click here to return to 'Tests of Knowledge'](#)



Crossword Puzzle - Answers

ACROSS

- | | |
|---------------------------------|-----------|
| 4. Product of protein decay | AMMONIA |
| 7. Soil on the move | EROSION |
| 8. Groundwater layer | AQUIFER |
| 9. pH less than 7 | ACID |
| 11. Power of the Hydrogen | pH |
| 13. Important in root growth | PHOSPHATE |
| 16. Particles of organic matter | DETRITUS |
| 17. Primary plant nutrient | NITRATE |

DOWN

- | | |
|-----------------------------------|-------------|
| 1. Buffers pH | ALKALINITY |
| 2. What coliform is | BACTERIA |
| 3. Measures calcium and magnesium | HARDNESS |
| 5. Urban runoff | STORMWATER |
| 6. What oxygen is in water | DISSOLVED |
| 10. Measured in °F and °C | TEMPERATURE |
| 11. Pollution from a pipe | POINTSOURCE |
| 12. pH greater than 7 | BASE |
| 14. Saltwater measure | SALINITY |
| 15. From living matter | ORGANIC |

Click here
to return to
'Tests of
Knowledge'



The End

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